

0040-4020(94)E0155-M

SYNTHESIS OF SAFRAMYCINS. IX. AN EFFICIENT SYNTHESIS OF THE ABC RING OF SAFRACINS.

Naoki Saito, Yasuko Obara, Tomoko Aihara, Shunji Harada, Yukiko Shida, and Arinori Kubo* Meiji College of Pharmacy, 1-35-23 Nozawa, Setagaya-ku, Tokyo 154, Japan

Abstract: 1,2,3,4,5,6-Hexahydro-10-hydroxy-1,5-imino-9-methoxy-3,8,11trimethyl-3-benzazocine 3b embodying all of the skeletal features of the ABC ring of safracins has been synthesized from compound 4a via a direct regioselective bromination, followed by the sequence reduction, metal-halogen interchange, and reaction of the organometallic intermediate with nitrobenzene. And the conversion of 3b to a quinone 21b is also described.

Sometime ago we became interested in safracins A 1a and B 1b, which were first isolated by the Yoshitomi Laboratories group from *Pseudomonas fluorescens* A2-2 in 1983,¹⁾ as attractive synthetic targets because they were plausible and important biogenetic intermediates of saframycins 2a-c.²⁾ We recently have reported preparation of the ABC ring model compound 3a of safracins involving nitration of 4a, followed by hydrogenation to an amine 5 and subsequent diazotization of the amine 5 in the usual manner which gave the phenol 3a.³⁾ A more efficient synthetic method was required, however, because the overall yield of 3a in this route was rather low and this sequence could not be applied for the total synthesis of 1a. This paper describes an efficient synthesis of 3b using aromatic hydroxylation of bromide 10b and the conversion of 3b to a quinone 21b which embodies all of the skeletal features of the "right half" of saframycin B 2b.⁴)



Results and Discussion

The challenge addressed was the introduction of a hydroxy group into the C-10 position⁵) of the hexahydro-1,5-imino-3-benzazocine skeleton by means of Baeyer-Villiger reaction (Scheme I). Numerous efforts for the direct formylation of $4a^{3}$) employing usual reagents (Cl₂CHOCH₃-TiCl₄, hexamethylenetetramine-trifluoroacetic acid, POCl₃-dimethylformamide (DMF)) were totally unsuccessful, only the starting material was recovered. Accordingly, the sequence of reactions was studied. Treatment of 4a with boron tribromide at -20°C for 1 h afforded the phenol 7 in 75% yield.³) This material was then converted into the aldehyde 8 according to the procedure of Pandit⁶) in 97% yield. Methylation of 8 with dimethyl sulfate in the presence of sodium hydride in DMF at room temperature for 1 h afforded 6 in 81% yield. To prevent the formation of any unwanted N-oxides⁷) during Baeyer-Villiger reaction, the Matsumoto protocol was used.⁸) Treating 6 and 30% hydrogen peroxide in methanol with sulfuric acid under reflux for 23 h gave the phenol 3a in 7% yield and the ester 9 in 59% yield.⁹)



The yield of this process was disappointingly low because of the occurrence of unwanted hydrogen migration and proved exceedingly troublesome. This prompted us to examine the introduction of a hydroxy group which required a different approach (Scheme II).

Wiriyachitra and Cava¹¹) reported some isoquinoline alkaloids can be hydroxylated through the corresponding aryl bromides by a method based on a reaction first noted by Buck and Köbrich¹⁰). Treatment of **4a** with bromine in aqueous trifluoroacetic acid at 80°C for 2 h gave the bromide **10a** in 86% yield. We were surprised to find that treatment of **10a** with *sec*-BuLi in THF at -78°C followed by oxidation with nitrobenzene gave the expected phenol **3a** in only 5% yield. The major products **11** and **4a** were obtained in 21% and 19% yields, respectively. The structure of **11** was supported by the ¹³C NMR spectrum, which showed a peak at δ 83.9 (singlet) assigned to the carbon at C-5 position. The ¹H NMR spectrum also showed two doublet signals at δ 2.86 and 3.00 assigned to the methylene proton at C-6 position. Obviously, an unfavorable lithium-hydrogen exchange at C-5 position had occurred. We assumed that the methine proton enjoyed resonance

stabilization from an adjacent amide carbonyl group. Furthermore, treatment of 4a with sec-BuLi in THF at -78° C followed by oxidation with nitrobenzene afforded 11 in 53% yield.¹²) The prohibition of undesired reaction would probably best be accomplished using the amine 10b which was obtained by reduction of 10a with lithium aluminum hydride in 77% yield. Treatment of 10b with sec-BuLi in THF at -78° C followed by oxidation with nitrobenzene gave the phenol (3b) in 34% yield along with 4b in 45% yield. After extensive investigation of the reaction conditions, the following procedure was found to be best in terms of product yield and reproducibility of the reaction: Treatment of 10b with 15 equiv of nitrobenzene in THF at -78° C in the presence of 20 equiv of sec-BuLi provided the phenols 3b and 12 in 53% and 3% yields, respectively (along with 24% of 4b).¹³



Next, we turned our attention to preparing the phenol 3a from a polymethoxyarene 15 which was the key intermediate of the ABC ring model of saframycins.^{4a)} Previously we reported that the partial demethylation of 15 with boron tribromide in dichloromethane at -78°C for 1 h afforded the phenol 17 (Scheme III).^{4b)} The preferential cleavage of the more hindered ether group has already been documented for boron tribromide,¹⁴⁾ and it thus seemed reasonable that the structure of the phenol would be 17. We are now convinced that the phenol has a structure of 16 and this was confirmed by X-ray crystallographic analysis. We then investigated the conversion of 16 into the phenol 3a. Acylation of 16 with trifluoromethanesulfonic anhydride gave the triflate 18 in 87% yield. Treatment of 18 with formic acid, triethylamine, palladium acetate, and 1,1-bis(dipenylphosphino)ferrocene (DPPF) in DMF at 60°C for 2 h¹⁵) afforded the deoxygated product 19 in 89% yield. The structure of 19 was supported by the ¹H NMR spectrum, and irradiation of the aromatic proton at δ 6.68 caused an nuclear Overhauser enhancement (NOE) of the methyl signal at δ 2.22 (4%). After numerous

efforts under various of conditions,¹⁶) the O-demethylation of 19 with sodium benzylselenoate was achieved. Treatment of 19 with sodium benzylselenoate which was prepared *in situ* by the reaction of excess sodium borohydride and dibenzyl diselenide¹⁸) in DMF under reflux for 2 h gave the phenols 3a and 20 in 28% and 17% yields (38% yield of 19 was recovered). Interestingly, treatment of 19 with boron tribromide in dichloromethane at -78°C afforded the single product 20 in 30% yield. Thus, we efficiently synthesized the phenol 3a from 15 which was the key intermediate of the ABC ring model of saframycins.



With the ABC ring models of safracins 3a and 3b in hand, we then turned our attention to the conversion of the phenols 3a,b into the *p*-quinones 21a,b as a model conversion of safracins to saframycins (Scheme IV). Treatment of 3a with bis(salicylidene)ethylene-diiminocobalt (II) (salcomine)¹⁹) in DMF at room temperature for 5 h afforded 21a in 64% yield. Similarly, salcomine oxidation of 3b provided 21b in 55% yield. These quinones 21a,b were identical in all respects with authentic samples.^{4a}



On the other hand, oxidation of the amine 5 with Fremy's salt gave the imine 22 in 97.5% yield. Treatment of 22 with aqueous hydrochloric acid gave the quinones 21a and 23 in 50% and 40% yields, respectively.²⁰)

In summary, we have succeeded in synthesizing the ABC ring system of safracins from 4a and 15. The results described herein are being applied to the total synthesis of safracin A (1a) which is currently being undertaken in our laboratory.

Experimental Section

All melting points were determined with a Yanagimoto micromelting point apparatus and are uncorrected. UV spectra were determined in methanol with a Hitachi 340 spectrometer. IR spectra were obtained with a Hitachi 260-10 spectrophotometer and ¹H-NMR spectra were recorded at 270MHz with a JEOL JNM-EX 270 spectrometer. ¹³C-NMR were recorded at 67.5MHz (multiplicity determined from off-resonance decoupled or INEPT spectra). NMR spectra were measured in CDCl₃, and chemical shifts were recorded in $\delta_{\rm H}$ values relative to internal (CH₃)₄Si standard. Mass spectra were recorded on a JMS-DX 302 mass spectrometer. Elemental analyses were obtained by a Perkin-Elmer Model 240B elemental analyzer. All reactions were conducted under an argon atmosphere. Dry solvents and reagents were obtained using standard procedures. Anhydrous sodium sulfate was used for drying organic solvent extracts, and removal of the solvent was done with a rotary evaporator and, finally, under high vacuum. Column chromatography was performed with E. Merck silica gel 60 (70-230 mesh).

10-Formyl-1,2,3,4,5,6-hexahydro-9-hydroxy-1,5-imino-3,8,11-trimethyl-4-oxo-3-

benzazocine (8). A suspension of the phenol (7³): 49.2 mg, 0.2 mmol) and hexamethylenetetramine (280.9 mg, 2 mmol) in trifluoroacetic acid (9 mL) was heated under reflux for 16 h. The reaction mixture was diluted with water (30 mL), made alkaline with powdered NaHCO3, and extracted with dichloromethane (20 mL x 3). The combined extracts were washed with water (20 mL), dried, and concentrated in vacuo to give the residue (63.2 mg). Chromatography on a silica gel (10 g) column with dichloromethane-methanol (50:1 - 40:1) afforded 8 (53.2 mg, 97.1 %) as a solid, which was recrystallized from methanol to give pale yellow needles: mp 202-203°C; IR (KBr) 3600-3300, 1635 cm⁻¹; UV λ_{max} (log ε) 220 (4.16), 271 (3.96), 354 (3.55) nm; ¹H NMR δ 2.21 (3H, s, ArCH3), 2.52 (3H, s, NCH3), 2.77 (1H, d, J = 17.2 Hz, 6-H β), 2.87 (3H, s, NCH3), 3.15 (1H, d, J = 11.6 Hz, 2-H β), 3.17 (1H, dd, J = 17.2, 6.3 Hz, 6-H α), 3.66 (1H, d, J = 6.3 Hz, 5-H), 4.10 (1H, dd, J = 11.6, 5.0 Hz, 2-H α), 4.56 (1H, d, J = 5.0 Hz, 1-H), 7.13 (1H, s, 7-H), 10.23 (1H, s, CHO), 12.26 (1H, s, OH); ¹³C NMR δ 15.0 (q, ArCH3), 27.0 (t, 6-C), 33.8 (q, NCH3), 40.1 (q, NCH3), 51.0 (d, 1-C), 55.1 (t, 2-C), 58.6 (d, 5-C), 115.4 (s), 123.2 (s), 126.9 (s), 134.3 (s), 139.2 (d, 7-C), 161.4 (s), 169.8 (s, CO), 192.9 (s, CHO); MS, *m*/z (relative intensity) 274 (M⁺, 22), 203 (19), 202 (100). Anal. Calcd for C15H18N2O3·1/10H2O: C, 65.25; H, 6.64; N, 10.15. Found: C, 65.12; H, 6.65; N, 10.00.

10-Formyl-1,2,3,4,5,6-hexahydro-1,5-imino-9-methoxy-3,8,11-trimethyl-4-oxo-3benzazocine (6). Sodium hydride (60% oil dispersion, washed with dry hexane three times, 28.8 mg, 1.2 mmol) was added to a stirred solution of 8 (164.4 mg, 0.6 mmol) in dry DMF (10 mL), and the resulting solution was stirred for 30 min at room temperature. Dimethyl sulfate (108.4 μ L, 1.2 mmol) in dry DMF (2 mL) was added, and the reaction mixture was stirred for 1 h at room temperature. The reaction mixture was concentrated in vacuo, and the residue was diluted with ethyl acetate (20 mL), and extracted with 1N HCl (20 mL x 3). The acidic aqueous layer was made alkaline with diluted NH4OH and extracted with chloroform (20 mL x 3). The combined extracts were washed with water (20 mL), dried, and concentrated in vacuo to give the residue (145.3 mg). Chromatography on a silica gel (15 g) column with dichloromethane-methanol (50:1) afforded 6 (139.5 mg, 80.7 %) as a solid, which was recrystallized from ethyl acetate to give colorless needles: mp 154-156°C; IR (KBr) 1685, 1655 cm⁻¹; UV λ_{max} (log ε) 210 (4.38), 259 (3.96), 314 (3.51) nm; ¹H NMR δ 2.31 (3H, s, ArCH3), 2.47 (3H, s, NCH3), 2.83 (3H, s, NCH3), 2.86 (1H, d, J = 17.2 Hz, 6-H β), 3.10 (1H, d, J = 12.5 Hz, 2-H β), 3.21 (1H, dd, J = 17.2, 6.9 Hz, 6-H α), 3.61 (1H, d, J = 6.9 Hz, 5-H), 3.85 (3H, s, OCH3), 4.05 (1H, dd, J = 12.5, 5.3 Hz, 2-H α), 4.72 (1H, d, J = 5.3 Hz, 1-H), 7.22 (1H, s, 7-H), 10.51 (1H, s, CHO); ¹³C NMR δ 15.4 (q, ArCH3), 28.2 (t, 6-C), 33.9 (q, NCH3), 40.2 (q, NCH3), 52.5 (d, 1-C), 54.0 (t, 2-C), 58.4 (d, 5-C), 63.2 (q, OCH3), 125.6 (s), 129.4 (s), 131.1 (s), 135.8 (s), 138.3 (d, 7-C), 163.0 (s), 170.0 (s, CO), 193.0 (s, CHO); MS, *m/z* (relative intensity) 288 (M⁺, 18), 217 (22), 216 (100). Anal. Calcd for C1₆H₂₀N₂O₃·1/10H₂O: C, 66.23; H, 7.02; N, 9.66. Found: C, 66.18; H, 6.96; N, 9.62.

Baeyer-Villiger oxidation of 6. A solution of 6 (28.8 mg, 0.1 mmol), concentrated H2SO4 (7 μ L) in methanol (1 mL) was cooled with ice-water, and 30% hydrogen peroxide (20 μ L) was added dropwise over 5 min. The reaction mixture was heated under reflux for 23 h. The reaction mixture was diluted with water (10 mL), made alkaline with 5% NaHCO3, and extracted with chloroform (10 mL x 3). The combined extracts were washed with water (10 mL), dried, and concentrated in vacuo to give the residue (29.2 mg). Chromatography on a silica gel (8 g) column with dichloromethane-methanol (100:1 - 50:1) afforded 9 (18.6 mg, 58.5 %) as a solid, which was recrystallized from ether to give colorless needles. Further elution with dichloromethane-methanol (40:1) gave 3a (1.8 mg, 6.5 %) as colorless prisms, mp 217-218.5°C, which were identical in all respects with the authentic sample described earlier.³)

1,2,3,4,5,6-Hexahydro-1,5-imino-9-methoxy-3,8,11-trimethyl-4-oxo-3-benzazocine-10carboxylic Acid Methyl Ester (9): mp 132-133.5°C; IR (KBr) 1725, 1650 cm⁻¹; UV λ_{max} (log ε) 212 (4.28), 272 (3.18), 314 (3.30) nm; ¹H NMR δ 2.26 (3H, s, ArCH3), 2.46 (3H, s, NCH3), 2.84 (3H, s, NCH3), 2.84 (1H, d, J = 17.2 Hz, 6-H β), 3.17 (1H, dd, J = 17.2, 6.6 Hz, 6-H α), 3.28 (1H, dd, J = 14.5, 4.0 Hz, 2-H β), 3.62 (1H, d, J = 6.6 Hz, 5-H), 3.76 (3H, s, OCH3), 3.86 (1H, d, J = 14.5 Hz, 2-H α), 3.89 (1H, d, J = 4.0 Hz, 1-H), 3.94 (3H, s, OCH3), 7.00 (1H, s, 7-H); ¹³C NMR δ 15.8 (q, ArCH3), 27.4 (t, 6-C), 33.8 (q, NCH3), 40.0 (q, NCH3), 52.4 (q, OCH3), 53.5 (d, 1-C), 54.0 (t, 2-C), 58.9 (d, 5-C), 61.6 (q, OCH3), 126.6 (s), 128.5 (s), 130.6 (s), 130.7 (s), 133.3 (d, 7-C), 154.3 (s), 168.5 (s, CO), 169.9 (s, NHCO); MS, *m*/z (relative intensity) 318 (M⁺, 19), 247 (21), 246 (100). Anal. Calcd for C17H20N2O4·1/10H2O: C, 63.77; H, 6.99; N, 8.75. Found: C, 63.76; H, 6.92; N, 8.65.

10-Bromo-1,2,3,4,5,6-hexahydro-1,5-imino-9-methoxy-3,8,11-trimethyl-4-oxo-3benzazocine (10a). A carbon tetrachloride solution of bromine (1.0 M, 4.5 mL, 4.5 mmol) was added to a stirred solution of 4a (780 mg, 3 mmol) in trifluoroacetic acid (5 mL) and water (2 mL) at 80°C for 10 min, and the mixutre was heated at 80°C for 2 h. The reaction mixture was diluted with ether (30 mL), and extracted with 1N HCl (20 mL x 3). The acidic aqueous layer was made alkaline with diluted NH4OH and extracted with chloroform (20 mL x 3). The combined extracts were washed with water (20 mL), dried, and concentrated in vacuo to give the residue (997.9 mg). Chromatography on a silica gel (50 g) column with dichloromethanemethanol (100:3) afforded 10a (874.5 mg, 85.9 %) as a solid, which was recrystallized from ethyl acetate-ether to give colorless prisms: mp 117.5-119°C; IR (KBr) 1650 cm⁻¹; UV λ_{max} (log ε) 220 (4.11), 273 (3.01), 282 (3.05) nm; ¹H NMR δ 2.30 (3H, s, ArCH₃), 2.48 (3H, s, NCH₃), 2.83 (1H, d, J = 17.2 Hz, 6-H β), 2.85 (3H, s, NCH₃), 3.19 (1H, dd, J = 17.2, 6.9 Hz, 6-H α), 3.26 (1H, d, J = 11.9 Hz, 2-H β), 3.63 (1H, d, J = 6.9 Hz, 5-H), 3.79 (3H, s, OCH₃), 3.92 (1H, dd, J = 11.9, 5.0 Hz, 2-H α), 4.20 (1H, d, J = 5.0 Hz, 1-H), 6.91 (1H, s, 7-H); ¹³C NMR δ 16.4 (q, ArCH₃), 27.6 (t, 6-C), 33.9 (q, NCH₃), 40.0 (q, NCH₃), 52.7 (t, 2-C), 55.9 (d, 1-C), 59.0 (d, 5-C), 60.1 (q, OCH₃), 118.5 (s), 130.3 (s), 130.6 (d, 7-C), 131.9 (s), 132.3 (s), 154.0 (s), 170.0 (s, CO); MS, *m*/z (relative intensity) 340 (M⁺+2, 25), 338 (M⁺, 27), 269 (18), 268 (99), 267 (20), 266 (100). Anal. Calcd for C₁₅H₁₉N₂O₂Br: C, 53.11; H, 5.65; N, 8.17. Found: C, 53.01; H, 5.59; N, 8.17.

Hydroxylation of 10a. A solution of the bromide (10a: 33.9 mg, 0.1 mmol) in dry THF (2.9 mL) was added to a solution of *sec*-BuLi (1.08 M, 463 μ L, 0.5 mmol) in cyclohexane and dry THF (1.1 mL) and the mixture was allowed to stir for 45 min at -78°C. To this mixture at the same temperature, nitrobenzene (83.4 μ l, 0.81 mmol) was added quickly. After being kept at the same temperature for 1 h, the solution was brought to room temperature over 1 h. The reaction mixture was diluted with ether (10 mL) and extracted with 1N HCl (10 mL x 3). The acidic aqueous layer was made alkaline with diluted NH4OH and extracted with chloroform (10 mL x 3). The combined extracts were washed with water (10 mL), dried, and concentrated in vacuo to give the residue (28.1 mg). This material was subjected to chromatography on preparative layer silica gel plates (Merck 5715, solvent 15:1 dichloromethane-methanol) to afford the phenol (3a: 1.5 mg, 5.4 %), the alcohol (11: 5.9 mg, 21.3 %), and 4a (4.8 mg, 18.5 %).

1.2.3.4.5.6-Hexahydro-5-hydroxy-1.5-imino-9-methoxy-3.8.11-trimethyl-4-oxo-3-A solution of 4a (78.0 mg, 0.3 mmol) in dry THF (3 mL) was added to a solution benzazocine (11). of sec-BuLi (1.08 M, 1.39 mL, 1.5 mmol) in cyclohexane and dry THF (3 mL) and the mixture was stirred for 45 min at -78°C. To this mixture at the same temperature, nitrobenzene (309 µl, 3.0 mmol) was added quickly. After being kept at the same temperature for 1 h, the solution was brought to room temperature over 1 h. The reaction mixture was diluted with ether (20 mL), and extracted with 1N HCl (20 mL x 3). The acidic aqueous layer was made alkaline with diluted NH4OH and extracted with chloroform (20 mL x 3). The combined extracts were washed with water (20 mL), dried, and concentrated in vacuo to give the residue (67.1 mg). This material was subjected to chromatography (silica gel, 10 g; elution with 50:1 dichloromethane-methanol) to give the alcohol 11 (43.6 mg, 52.7 %) as a solid, which was recrystallized from methanol to give colorless needles: mp 258-260°C; IR (KBr) 3750-3350, 1670, 1630 cm⁻¹; UV λ_{max} (log ϵ) 220 (3.97), 230 (3.71), 278 (3.44), 286 (3.44) nm; ¹H NMR δ 2.18 (3H, s, ArCH₃), 2.45 (3H, s, NCH₃), 2.83 (3H, s, NCH₃), 2.86 (1H, d, J = 17.2 Hz, 6-H β), 3.00 (1H, d, J = 17.2 Hz, 6-H α), 3.09 (1H, d, J = 10.9 Hz, 2-H β), 3.82 (3H, s, OCH₃), 3.99 (1H, dd, J = 10.9, 4.9 Hz, 2-H α), 4.05 (1H, d, J = 4.9 Hz, 1-H), 6.56 (1H, s, 7-H), 6.84 (1H, s, 10-H); ¹³C NMR δ 15.9 (q, ArCH₃), 32.4 (t, 6-C), 34.6 (q, NCH₃), 34.9 (q, NCH₃), 55.4 (q, OCH₃), 56.0 (t, 2-C), 59.7 (d, 1-C), 83.9 (s, 5-C), 108.1 (d, 10-C), 124.5 (s), 126.5 (s), 130.7 (d, 7-C), 130.8 (s), 156.7 (s), 170.9 (s, CO); MS, m/z (relative intensity) 276 (M⁺, 15), 205 (25), 204 (100). Anal. Calcd for C15H20N2O3: C, 65.19; H, 7.30; N, 10.14. Found: C, 64.99; H, 7.30; N, 10.16.

10-Bromo-1,2,3,4,5,6-hexahydro-1,5-imino-9-methoxy-3,8,11-trimethyl-3-benzazocine (10b). Lithium aluminum hydride (151.6 mg, 2.15 mmol) was added to a stirred solution of 10a (729.4 mg, 2.15 mmol) in dry THF (40 mL) at 0°C, and then stirring was continued at 0°C for 1 h. After being quenched at 0°C by the addition of water, the mixture was filtered and the filter cake was carefully washed with chloroform (100 mL). The combined filtrates were concentrated in vacuo. The residue was diluted with ethyl acetate (50 mL), and extracted with 1N HCl (30 mL x 3). The acidic aqueous layer was made alkaline with diluted NH4OH and extracted with chloroform (20 mL x 3). The combined extracts were washed with water (20 mL), dried, and concentrated in vacuo to give the residue (717.1 mg). Chromatography on a silica gel (40 g) column with ethyl acetate-methanol (40:1) afforded 10b (502.4 mg, 77.2 %) as a solid, which was recrystallized from ether to give colorless needles: mp 60-61°C; IR (KBr) 2990, 2970, 2940, 2905, 2890, 2840, 2790, 2730, 1655, 1470, 1450, 1440, 1375, 1365, 1310, 1285, 1235, 1200, 1175, 1130, 1080, 1055, 1025, 1000, 985, 950 cm⁻¹; UV λ_{max} (log ε) 220 (4.10), 240sh (3.83), 273 (3.04), 282 (3.10) nm; ¹H NMR δ 2.13 (3H, s, NCH₃), 2.28 (3H, s, ArCH₃), 2.30 (3H, s, NCH₃), 2.37 (1H, dd, J = 10.6, 3.0 Hz, 2-H α), 2.45 $(1H, dd, J = 10.6, 2.3 Hz, 4-H\alpha), 2.54 (1H, dd, J = 20.5, 3.6 Hz, 6-H\beta), 2.75 (1H, ddd, J = 10.6, 1.0, 1.0)$ Hz, 4-H β), 2.89 (1H, ddd, J = 10.6, 1.0, 1.0 Hz, 2-H β), 3.09 (1H, dd, J = 20.5, 7.6 Hz, 6-H α), 3.09 (1H, dddd, J = 7.6, 3.6, 2.3, 1.0 Hz, 5-H), 3.79 (3H, s, OCH3), 4.03 (1H, br s, 1-H), 6.87 (1H, s, 7-H); ¹³C NMR & 16.4 (q, ArCH3), 26.5 (t, 6-C), 41.1 (q, NCH3), 46.5 (q, NCH3), 52.9 (d, 5-C), 58.5 (d, 1-C), 58.6 (t, 2-C), 60.0 (q, OCH3), 62.6 (t, 4-C), 118.4 (s), 129.4 (d, 7-C), 130.4 (s), 132.7 (s), 133.7 (s), 152.9 (s); MS, m/z (relative intensity) 326 (M⁺+2, 5), 324 (M⁺, 5), 269 (18), 268 (99), 267 (15), 266 (100). Anal. Calcd for C15H21N2OBr 1/5H2O: C, 54.79; H, 6.56; N, 8.52. Found: C, 54.80; H, 6.57; N, 8.41.

Hydroxylation of 10b. (Method A): A solution of the bromide (10b: 65.1 mg, 0.2 mmol) in dry THF (5 mL) was added to a solution of sec-BuLi (1.08 M, 1.93 mL, 2.08 mmol) in cyclohexane and dry THF (5 mL) and the mixture was allowed to stir for 45 min at -78°C. To this mixture at the same temperature, nitrobenzene (164.7 μ l, 1.6 mmol) was added quickly. After being kept at the same temperature for 1 h, the solution was brought to room temperature over 1 h. The reaction mixture was diluted with ether (20 mL), and extracted with 1N HCl (20 mL x 3). The acidic aqueous layer was made alkaline with diluted NH4OH and extracted with chloroform (20 mL x 3). The combined extracts were washed with water (10 mL), dried, and concentrated in vacuo to give the residue (111.4 mg). Chromatography on a silica gel (10 g) column with dichloromethane-methanol (100:3) afforded 4b (21.9 mg, 44.5 %) as a solid, which was recrystallized from ethyl acetate-ether to give colorless needles. Further elution with dichloromethane-methanol (20:1 - 10:1) gave 3b (17.7 mg, 33.8 %) as a solid, which was recrystallized from methanol to give colorless prisms. (Method B). A solution of sec-BuLi (1.08 M, 3.70 mL, 4.0 mmol) in cyclohexane was added quickly to a stirred solution of 10b (65.1 mg, 0.2 mmol) and nitrobenzene (309 µL, 3.0 mmol) in dry THF (10 mL) at -78°C. After being kept at the same temperature for 4 h, the solution was brought to -30°C for 30 min. The reaction mixture was diluted with ether (20 mL), and extracted with 1N HCl (20 mL x 3). The acidic aqueous layer made alkaline with diluted NH4OH and extracted with chloroform (20 mL x 3). The combined extracts were washed with water (10 mL), dried, and concentrated in vacuo to give the residue (133.1 mg). Chromatography on a silica gel (10 g) column with dichloromethane-methanol (50:1) afforded the starting material 10b (15.0 mg, 23.0 %) and dichromethne-methanol (100:3) afforded 4b (12.0 mg, 24.4 %) as a solid, which was recrystallized from ethyl acetate-ether to give colorless needles. Further elution with dichloromethane-methanol (20:1 - 10:1) gave the residue (32.0 mg), which showed two major spots on TLC (Rf 0.11 and 0.09, 9:1 chloroform-methanol). This material was subjected to chromatography on a silica gel, 6 g) column with ethyl acetate-methanol (20:1) and afforded 12 (2.4 mg, 3.4 %), and with ethyl acetate-methanol (10:1 - 3:1) afforded 3b (27.5 mg, 52.5 %) as a solid, which was recrystallized from acetone to give colorless prisms.

(3b), mp 184.5-185°C: IR (KBr) 3000, 2930, 2880, 2800, 2615, 1585, 1495, 1465, 1445, 1435, 1415, 1375, 1350, 1325, 1295, 1255, 1230, 1175, 1130, 1080, 1060, 1045, 1025, 1000, 982, 945, 860, 845, 835, 785, 745 cm⁻¹; UV λ_{max} (log ϵ) 222 (3.98), 272 (3.89), 280 (3.93) nm; ¹H NMR δ 2.14 (3H, s, NCH₃), 2.24 (3H, s, ArCH₃), 2.33 (3H, s, NCH₃), 2.41 (1H, dd, J = 10.6, 3.0 Hz, 2-H α), 2.47 (1H, dd, J = 10.9, 2.9 Hz, 4-Ha), 2.54 (1H, d, J = 17.2 Hz, 6-Hb), 2.76 (1H, d, J = 10.9 Hz, 4-Hb), 2.86 (1H, ddd, J = 10.6, 2.0, 2.0 Hz, 2-H β), 3.07 (1H, dd, J = 17.2, 7.6 Hz, 6-H α), 3.10 (1H, ddd, J = 7.6, 2.0, 1.0 Hz, 5-H), 3.76 (3H, s, OCH3), 4.04 (1H, br s, 1-H), 6.47 (1H, s, 7-H); ¹³C NMR δ 15.8 (q, ArCH3), 26.4 (t, 6-C), 41.0 (q, NCH3), 46.6 (q, NCH3), 52.9 (d, 5-C), 53.0 (d, 1-C), 59.2 (t, 2-C), 60.7 (q, OCH3), 62.7 (t, 4-C), 120.3 (s), 121.0 (d, 7-C), 128.0 (s), 131.6 (s), 142.8 (s), 145.4 (s); MS, m/z (relative intensity) 262 (M⁺, 10), 205 (15), 204 (100), 189 (13), 173 (6). Anal. Calcd for C15H22N2O2: C, 68.67; H, 8.45; N, 10.68. Found: C, 68.60; H, 8.70; N, 10.67.

1,2,3,4,5,6-Hexahydro-10-hydroxy-1,5-imino-9-methoxy-3,8,11-trimethyl-7-phenylaminomp 212-214°C; IR (KBr) 3310, 2950, 2910, 2805, 2602, 1515, 1500, 1460, 3-benzazocine (12). 1438, 1415, 1385, 1370, 1350, 1322, 1302, 1265, 1255, 1238, 1178, 1118, 1085, 1060, 1045, 1012, 998, 975, 950, 860, 790, 745 cm⁻¹; UV λ_{max} (log ϵ) 246 (4.10), 268sh (3.96), 302sh (3.40) nm; ¹H NMR δ 2.13 $(3H, s, ArCH_3)$, 2.15 $(3H, s, NCH_3)$, 2.30 $(3H, s, NCH_3)$, 2.43 $(1H, dd, J = 10.6, 3.0 Hz, 2-H\alpha)$, 2.44 $(1H, dd, J = 10.9, 3.6 Hz, 4-H\alpha), 2.47 (1H, d, J = 18.1 Hz, 6-H\beta), 2.71 (1H, ddd, J = 10.9, 1.0, 1.0 Hz, 1.0 Hz)$ 4-HB), 2.83 (1H, dd, J = 18.1, 7.6 Hz, 6-HB), 2.89 (1H, ddd, J = 10.6, 1.0, 1.0 Hz, 2-HB), 3.07 (1H, br d, 5-H), 3.78 (3H, s, OCH3), 4.12 (1H, br s, 1-H), 5.02 (1H, br s, NH), 6.46 (2H, dd, J = 8.6, 1.0 Hz, ArH x 2), 6.71 (1H, t like, ArH), 7.13 (2H, dd, J = 8.6, 7.3 Hz, ArH x 2); ¹³C NMR δ 11.4 (q, ArCH₃), 23.8 (t, 6-C), 41.0 (q, NCH3), 46.5 (q, NCH3), 52.7 (d, 5-C), 53.0 (d, 1-C), 59.1 (t, 2-C), 60.9 (q, OCH3), 62.5 (t, 4-C), 112.6 (d), 117.6 (s), 120.5 (d), 127.7 (s), 128.8 (s), 129.4 (d), 131.0 (s), 143.2 (s), 144.2 (s), 146.9 (s); MS, m/z (relative intensity) 353 (M⁺, 15), 296 (23), 295 (100), 280 (14). Anal. Calcd for C₂₁H₂₇N₃O₂: C, 71.36; H, 7.70; N, 11.89. Found: C, 71.36; H, 7.97; N, 11.64.

1,2,3,4,5,6-Hexahydro-1,5-imino-9-methoxy-3,8,11-trimethyl-3-benzazocine (4b). A stirred solution of 4a (260 mg, 1.0 mmol) in dry THF (30 mL) was cooled with ice-water, a THF solution of aluminum hydride (0.5 M, 12 mL, 6 mmol) was added dropwise over 10 min, and then stirring was continued at 0°C for 1 h. After being guenched by the addition of methanol, the reaction mixture was concentrated in vacuo. The residue (302.1 mg) was subjected to chromatography (silica gel, 16 g; elution with 100:3 - 20:1 dichloromethane-methanol) to give 4b (211.5 mg, 86.0 %) as a solid, which was recrystallized from ethyl acetate-ether to give colorless prisms: mp 88-89.5°C; IR (KBr) 3010, 2970, 2940, 2900, 2830, 2790, 2760, 1615, 1505, 1455, 1415, 1375, 1360, 1350, 1330, 1310, 1295, 1275, 1260, 1230, 1215, 1170, 1150, 1105, 1080, 1065, 1050, 1035, 1010, 1000, 950, 920, 880, 855, 845, 820, 785, 760, 740 cm⁻¹; UV λ_{max} (log ϵ) 220 (3.94), 238sh (3.75), 282 (3.49), 288 (3.48) nm; ¹H NMR δ 2.13 (3H, s, NCH3), 2.16 (3H, s, ArCH₃), 2.34 (3H, s, NCH₃), 2.44 (2H, dd, J = 10.6, 3.0 Hz, 2-H α and 4-H α), 2.54 (1H, d, J = 16.8 Hz, 6-HB), 2.75-2.81 (2H, m, 4-HB and 2-HB), 3.05 (1H, dd, J = 16.8, 7.6 Hz, 6-H α), 3.09 (1H, br s, 5-H), 3.60 (1H, br s, 1-H), 3.79 (3H, s, OCH₃), 6.47 (1H, s, 7-H), 6.85 (1H, s, 10-H); 1^{3} C NMR δ 16.0 (q, ArCH₃), 26.2 (t, 6-C), 41.2 (q, NCH₃), 46.6 (q, NCH₃), 53.3 (d, 5-C), 55.4 (q, OCH₃), 59.1 (d, 1-C), 61.5 (t, 2-C), 62.9 (t, 4-C), 108.6 (d, 10-C), 124.7 (s), 126.6 (s), 129.8 (d, 7-C), 134.0 (s), 155.7 (s); MS,

1,2,3,4,5,6-Hexahydro-10-hydroxy-1,5-imino-9-methoxy-3,8,11-trimethyl-3-benzazocine

m/z (relative intensity) 246 (M⁺, 10), 189 (15), 188 (100), 173 (7). Anal. Calcd for C₁₅H₂₂N₂O: C, 73.13; H, 9.00; N, 11.37. Found: C, 72.94; H, 9.11; N, 11.35.

10-Acetoxy-1,2,3,4,5,6-hexahydro-1,5-imino-9-methoxy-3,8,11-trimethyl-3-benzazo-Acetic anhydride (0.2 mL) was added to a solution of 3b (34.7 mg, 0.132 mmol) in dry pyridine cine (13). (0.5 mL), and the mixture was left to stand at room temperature for 1 h. After being diluted with water (5 mL). the mixture was extracted with chloroform (10 mL x 3). The combined extracts were washed with 5% NaHCO3 (10 mL), dried, and concentrated in vacuo to give the residue (28.2 mg). This material was subjected to chromatography on a silica gel (6 g) column with dichloromethane-methanol (100:1 - 50:1) to afford 13 (20.5 mg, 50.9 %) as a solid, which was recrystallized from ether-hexane to give colorless needles: mp 117-119°C; IR (KBr) 1765 cm⁻¹; UV λ_{max} (log ε) 220 (4.01), 270 (3.96), 278 (3.96) nm; ¹H NMR δ 2.13 (3H, s, NCH3), 2.25 (3H, s, ArCH3), 2.28 (3H, s, NCH3), 2.33 (3H, s, COCH3), 2.39 (1H, dd, J = 10.6, 3.0 Hz, 2-H α), 2.45 (1H, dd, J = 10.6, 2.0 Hz, 4-Ha), 2.57 (1H, d, J = 16.8 Hz, 6-HB), 2.72 (1H, dd, J = 10.6, 2.0 Hz, 2-HB), 2.76 (1H, d, J = 10.6, Hz, 4-HB), 3.08 (1H, dd, J = 16.8, 7.6 Hz, 6-H α), 3.11 (1H, br, 5-H), 3.70 (1H, br s, 1-H), 3.72 (3H, s, OCH₃), 6.82 (1H, s, 7-H); ¹³C NMR δ 15.8 (q, ArCH₃), 20.5 (q, COCH₃), 26.1 (t, 6-C), 40.9 (q, NCH3), 46.4 (q, NCH3), 52.7 (d, 5-C), 53.7 (d, 1-C), 59.6 (t, 2-C), 60.4 (q, OCH3), 62.6 (t, 4-C), 126.6 (s), 127.6 (d, 7-C), 129.9 (s), 131.3 (s), 141.3 (s), 147.6 (s), 169.1 (CO); MS, m/z (relative intensity) 304 (M⁺, 13), 247 (20), 246 (100), 205 (10), 204 (68). Anal. Calcd for C17H24N2O3: C, 67.08; H, 7.95; N, 9.20. Found: C, 66.85; H, 8.29; N, 9.09.

10-Acetoxy-1,2,3,4,5,6-hexahydro-1,5-imino-9-methoxy-3,8,11-trimethyl-7-phenyl-Acetic anhydride (0.2 mL) was added to a solution of 12 (12.8 mg, amino-3-benzazocine (14). 0.0363 mmol) in dry pyridine (0.5 mL), and the mixture was left to stand at room temperature for 1 h. After being diluted with water (5 mL), the mixture was extracted with chloroform (10 mL x 3). The combined extracts were washed with 5% NaHCO3 (10 mL), dried, and concentrated in vacuo to give the residue (19.2 mg). This material was subjected to chromatography on a silica gel (6 g) column with dichloromethane-methanol (100:1 -50:1) to afford 14 (12.9 mg, 90.1 %) as a solid, which was recrystallized from ether-hexane to give colorless needles: mp 151.5-153°C; IR (KBr) 3380, 1762, 1605 cm⁻¹; UV λ_{max} (log ε) 241 (4.11), 276 (3.91) nm; ¹H NMR & 2.13 (6H, s, NCH3 and ArCH3), 2.26 (3H, s, NCH3), 2.36 (3H, s, COCH3), 2.38-2.44 (2H, m, 4-H α and 2-H α), 2.47 (1H, d, J = 18.2 Hz, 6-H β), 2.69 (1H, ddd, J = 10.6, 1.0, 1.0 Hz, 2-H β), 2.73 (1H, ddd, J = 10.0, 1.0, 1.0 Hz, 4-H β), 2.84 (1H, dd, J = 18.2, 7.6 Hz, 6-H β), 3.07 (1H, br d, 5-H), 3.74 (1H, dd, J = 18.2, 7.6 Hz, 6-H β), 3.74 (1H, dd, J = 18.2, 7.6 Hz, 6-H β), 3.74 (1H, dd, J = 18.2, 7.6 Hz, 8-H β), 3.74 (1H, dd, J = 18.2, 7.6 Hz, 8-H β), 3.74 (1H, dd, J = 18.2, 7.6 Hz, 8-H β), 3.74 (1H, dd, J = 18.2, 7.6 Hz, 8-H β), 3.74 (1H, dd, J = 18.2, 7.6 Hz, 8-H β), 3.74 (1H, dd, J = 18.2, 7.6 Hz, 8-H β), 3.74 (1H, dd, J = 18.2, 7.6 Hz, 8-H β), 3.74 (1H, dd, J s like, 1-H), 3.74 (3H, s, OCH3), 5.11 (1H, br s, NH), 6.52 (2H, d, J = 8.6, 1.0 Hz, ArH x 2), 6.71 (1H, t, J = 8.3 Hz, ArH), 7.16 (2H, t, J = 8.3 Hz, ArH x 2); ¹³C NMR δ 11.5 (q, ArCH₃), 20.6 (q, COCH₃), 23.7 (t, 6-C), 40.9 (q, NCH3), 46.3 (q, NCH3), 52.6 (d, 5-C), 53.8 (d, 1-C), 59.5 (t, 2-C), 60.7 (q, OCH3), 62.5 (t, 4-C), 113.2 (d), 118.2 (d), 127.0 (s), 128.8 (s), 129.3 (s), 130.2 (s), 134.9 (s), 139.7 (s), 145.9 (s), 148.2 (s), 169.2 (CO); MS, m/z (relative intensity) 395 (M⁺, 15), 338 (25), 337 (100), 296 (11), 295 (53). Anal. Calcd for C23H29N3O3 1/5H2O: C, 69.22; H, 7.43; N, 10.53. Found: C, 69.39; H, 7.74; N, 10.23.

X-ray structure determination of 16. Crystals of **16** (C1₆H₂₂N₂O₄)^{4a}) belong to triclinic space group P1, with cell constants a = 8.6585 (179) Å, b = 13.1325 (211) Å, c = 7.5576 (116) Å, Z = 2, $d_c = 1.314$ cm⁻¹. X-ray intensities were measured at 296 K with an AFC-5 (Rikagu Denki) type diffractometer using graphite-monochromated Cu K α radiation, ω -2 θ scan mode, $3^{\circ} \leq 2\theta \leq 120^{\circ}$, number of reflexcions measured 2228, number of reflections with F (o) > 3 δ (Fo) 2138. The structure was solved by direct methods.

Refinements were done by a local block-diagonal version of UNICS III system (Open program Tokyo University). Hydrogen atoms were found from the difference in Fourier syntheses. The final R factor was 9.83 %. The drawing of the molecule was made by ORTEP.

1.2.3.4.5.6-Hexahvdro-7-hvdroxy-1.5-imino-9,10-dimethoxy-3.8,11-trimethyl-4-oxo-3-benzazocine Trifluoromethanesulfonate (18). Trifluoromethanesulfonic anhydride (53.2 µL, 32.0 mmol) was added to a stirred solution of 16 (54.4 mg, 0.18 mmol) and triethylamine (88.8 µL, 65.0 mmol) in dry dichloromethane (3 mL) at -20°C. After being kept at the same temperature for 30 min, the reaction mixture was poured into 5% NaHCO3 and the phases were separated. The aqueous layer was extracted with dichloromethane (10 mL x 2). The combined extracts were washed with water (20 mL), dried, and concentrated in vacuo to give the residue (94.4 mg). Chromatography on a silica gel (10 g) column with dichloromethanemethanol (100:1 - 80:1) afforded 18 (67.7 mg, 87.0 %) as a solid, which was recrystallized from acetone to give colorless prisms: mp 138-140°C; IR (KBr) 1665 cm⁻¹; UV λ_{max} (log ϵ) 221 (4.38) nm; ¹H NMR δ 2.27 (3H, s, ArCH₃), 2.50 (3H, s, NCH₃), 2.85 (3H, s, NCH₃), 2.92 (1H, d, J = 17.8 Hz, 6-Hβ), 3.07 (1H, d, J = 11.9 Hz, 2-HB), 3.17 (1H, dd, J = 17.8, 6.6 Hz, 6-H α), 3.68 (1H, d, J = 6.6 Hz, 5-H), 3.81 (3H, s, OCH3), 3.84 (3H, s, OCH3), 3.96 (1H, dd, J = 11.9, 4.6 Hz, 2-H α), 4.15 (1H, d, J = 4.6 Hz, 1-H); ¹³C NMR & 10.7 (q, ArCH3), 23.5 (t, 6-C), 34.0 (q, NCH3), 40.0 (q, NCH3), 51.3 (d, 1-C), 53.4 (t, 2-C), 58.1 (d, 5-C), 60.3 (q, OCH3), 60.4 (q, OCH3), 123.3 (s), 126.5 (s), 141.0 (s), 149.5 (s), 149.9 (s), 169.4 (s, CO); MS, m/z (relative intensity) 438 (M⁺, 18), 367 (21), 366 (56), 306 (22), 305 (100), 234 (25), 233 (41), 218 (35). Anal. Calcd for C17H21N2O6F3S: C, 46.57; H, 4.83; N, 6.39. Found: C, 46.71; H, 4.85; N, 6.42.

1.2.3.4.5.6-Hexahydro-1.5-imino-9.10-dimethoxy-3.8.11-trimethyl-4-oxo-3-benzazocine (19). Formic acid (71 µL, 0.64 mmol) was added to a mixture of 18 (68.6 mg, 0.16 mmol), triethylamine (130 µL, 0.96 mmol), palladium acetate (7.0 mg, 0.03 mmol), and 1,1-bis(dipenylphosphino)ferrocene (DPPF: 35.0 mg, 0.064 mmol) in dry DMF (1.5 mL) at 0°C for 10 min. The reaction mixture was stirred for 1 h at room temperature, and then heated at 60°C for 2 h. The reaction mixture was poured onto benzene (20 mL), and extracted with 1N HCl (20 mL x 3). The acidic aqueous layer was made alkaline with diluted NH4OH and extracted with chloroform (20 mL x 3). The combined extracts were washed with water (20 mL), dried, and concentrated in vacuo to give the residue (84.0 mg). Chromatography on a silica gel (10 g) column with ethyl acetate-methanol (30:1 - 10:1) afforded 19 (40.2 mg, 89.3 %) as a solid, which was recrystallized from ethyl acetate to give colorless prisms: mp 102-104°C; IR (KBr) 1650 cm⁻¹; UV λ_{max} (log ϵ) 220 (4.32), 269 (3.88), 276 (3.85) nm; ¹H NMR δ 2.22 (3H, s, ArCH₃), 2.50 (3H, s, NCH₃), 2.81 (1H, d, J = 17.2 Hz, 6-HB), 2.85 (3H, s, NCH3), 3.08 (1H, dd, J = 11.6, 1.0 Hz, 2-HB), 3.14 (1H, dd, J = 17.2, 6.9 Hz, 6-H α), 3.61 (1H, d, J = 6.9 Hz, 5-H), 3.79 (3H, s, OCH₃), 3.91 (3H, s, OCH₃), 3.95 (1H, dd, J =11.6, 4.9 Hz, 2-Ha), 4.13 (1H, d, J = 4.9 Hz, 1-H), 6.68 (1H, s, 7-H); ¹³C NMR δ 15.8 (q, ArCH₃), 27.6 (t, 6-C), 34.0 (g, NCH3), 40.1 (g, NCH3), 51.2 (d, 1-C), 53.7 (t, 2-C), 59.2 (d, 5-C), 59.9 (g, OCH3), 60.2 (q, OCH₃), 125.6 (d, 7-C), 126.0 (s), 128.4 (s), 131.5 (s), 149.1 (s), 149.6 (s), 170.2 (s, CO); MS, m/z (relative intensity) 290 (M⁺, 27), 219 (21), 218 (100). Anal. Calcd for C₁₆H₂₂N₂O₃: C, 66.18; H, 7.64; N, 9.54. Found: C, 66.02; H, 7.67; N, 9.54.

Partial Demethylation of 19. 1) Using Boron Tribromide: A dichloromethane solution of boron tribromide (1.0 M, 0.3 mL, 0.3 mmol) was added to a stirred solution of **19** (29.0 mg, 0.1 mmol) in dry dichloromethane (3 mL) at -78°C. After being kept at the same temperature for 2 h, and then at 0°C for 1 h, the

reaction mixture was poured into 5% NaHCO3 (10 mL) and the phases were separated. The aqueous layer was extracted with dichloromethane (10 mL x 2). The combined extracts were washed with water, dried, and concentrated in vacuo to give the residue (16.8 mg). Chromatography on a silica gel (5 g) column with dichloromethane-methanol (50:1 - 30:1) afforded 20 (5.8 mg, 29.6 %) as a solid, which was recrystallized from acetone to give colorless prisms: mp 178-180°C; IR (KBr) 3600-3040, 1640, 1620 cm⁻¹; UV λ_{max} (log ϵ) 218 (4.39), 276 (4.17), 282 (4.18) nm; ¹H NMR δ 2.21 (3H, s, ArCH₃), 2.54 (3H, s, NCH₃), 2.80 (1H, d, J = 17.1 Hz, 6-H β), 2.83 (3H, s, NCH₃), 3.11 (1H, dd, J = 17.1, 6.6 Hz, 6-H α), 3.12 (1H, d, J = 11.7 Hz, 2-H β), 3.64 (1H, d, J = 6.6 Hz, 5-H), 3.82 (3H, s, OCH₃), 3.98 (1H, dd, J = 11.7, 4.9 Hz, 2-H α), 4.10 (1H, d, J = 4.9 Hz, 1-H), 5.45 (1H, br s, OH), 6.67 (1H, s, 7-H); ^{13}C NMR δ 15.6 (q, ArCH₃), 27.2 (t, 6-C), 34.0 (q, NCH3), 40.0 (q, NCH3), 51.4 (d, 1-C), 53.6 (t, 2-C), 59.2 (d, 5-C), 60.8 (q, OCH3), 124.1 (s), 124.7 (s), 125.2 (s), 126.5 (d, 7-C), 143.8 (s), 145.5 (s), 169.8 (s, CO); MS, m/z (relative intensity) 276 (M⁺, 29), 205 (20), 204 (100), 189 (24), high-resolution MS calcd for C15H20N2O3 276.1473, found 276.1473. 2) Using Sodium Benzylselenoate: Sodium borohydride (20 mg, 0.5277 mmol) was added to a stirred solution of dibenzyl diselenide (11.0 mg, 0.068 mmol) in dry DMF (1 mL) at room temperature. After being kept at the same temperature for 15 min, a solution of 19 (29.0 mg, 0.1 mmol) in dry DMF (1 mL) was added quickly. The reaction mixture was heated under reflux for 2 h, and then concentrated in vacuo. The residue was diluted with benzene (20 mL), and extracted with 1N HCl (20 mL x 3). The acidic aqueous layer was made alkaline with NH4OH and extracted with chloroform (20 mL x 3). The combined extracts were washed with water (20 mL), dried, and concentrated in vacuo to give the residue. This material was subjected to chromatography on preparative layer silica gel plates (Merck 5715, solvent 5:4 dichloromethane-acetone) to afford **3a** (7.6 mg, 27.5 %), **20** (3.4 mg, 16.8 %), and **19** (11.1 mg, 38.3 % recovery).

1,2,3,4,5,6,7,10-Octahydro-1,5-imino-9-methoxy-3,8,11-trimethyl-4,7,10-trioxo-3benzazocine (21a). 1) Oxidation of the phenol (3a) with salcomine: Bis(salicylidene)ethylenediiminocobalt(II) (17.0 mg, 0.052 mmol) was added to a stirred solution of 3a (27.4 mg, 0.0993 mmol) in dry DMF (3 mL). The dark suspension was stirred under an oxygen atmosphere for 5 h at room temperature. The mixture was filtered through cellulose powder and the filter cake was carefully washed with ethyl acetate. The combined filtrates were washed with water (20 mL), dried, and concentrated in vacuo to give the residue (33.0 mg). The material was then subjected to chromatography on silica gel (8 g) with dichloromethane-methanol (50:1) to afford 21a (18.4 mg, 63.9 %) as a solid. This material was recrystallized from ethyl acetate-ether to give pale yellow prisms: mp 150-152 °C, whose spectra were identical with those of the authentic sample described earlier.^{4a)} 2) Oxidation of the amine (5) with Fremy's salts: A solution of the amine (5³): 55.0 mg, 0.2 mmol) in methanol (0.5 mL) was added quickly to a stirred solution of potassium nitrosodisulfonate (223.6 mg, 0.5 mmol) and aqueous potassium dihydrogen phosphate (0.02 M, 7.5 mL). The mixture was stirred for 1 h, poured into water (10 mL), and extracted with chloroform (10 mL x 3). The combined extracts were washed with water, dried, and concentrated in vacuo to give the imine (22: 56.4 mg, 97.5 %) as a solid, which was used in the next step without further purification. An analytical sample was obtained by recrystallization from ethyl acetate: mp 171-173°C; IR (KBr) 3230, 1645, 1620 cm⁻¹; UV λ_{max} (log ε) 268 (4.22), 320 (3.78) nm; ¹H NMR δ 2.00 (3H, s, quinone CH3), 2.48 (3H, s, NCH3), 2.76 (1H, d, J = 20.1 Hz, 6-H β), 2.85 (1H, dd, J = 20.1, 4.0 Hz, 6-H α), 2.89 (3H, s, NCH₃), 3.25 (1H, dd, J = 12.2, 1.0 Hz, 2-H β), 3.65 (1H, d, J = 4.0 Hz, 5-H), 3.86 (3H, s, OCH₃), 3.94 (1H, dd, J = 12.2, 5.3 Hz, 2-H α), 4.34 (1H, d, J = 5.3 Hz, 1-H), 11.08 (1H, s, NH); MS, m/z (relative intensity) 289 (M⁺, 100), 274 (30), 231 (20), 203 (86). Anal. Calcd for C15H19N3O3: C, 62.07; H, 6.61; N, 14.52. Found: C, 62.07; H, 6.61; N, 14.38.

A solution of the crude 22 (56.4 mg, 0.195 mmol) in 2N HCl (4 mL) was heated at 60°C for 1 h. After being diluted with water (10 mL), the pH was brought to 8-9 with solid NaHCO3 powder, and extracted with chloroform (20 mL x 3). The combined extracts were washed with water (10 mL), dried, and concentrated in vacuo to give the residue (35.6 mg). Chromatography on a silica gel (8 g) column with dichloromethanemethanol (100:1 - 80:1) afforded 21a (29.2 mg, 50.3 %: 2 steps) as a solid, which was identical in all respects with 21a prepared as above. The pH of the basic aqueous layer was brought to 5-6 with acetic acid, and extracted with chloroform (20 mL x 3). The combined extracts were washed with water, dried, and concentrated in vacuo to give 23 (22.2 mg, 40.2 %) as a solid, which was recrystallized from methanol to give pale yellow needles:

1,2,3,4,5,6,7,10-Octahydro-9-hydroxy-1,5-imino-3,8,11-trimethyl-4,7,10-trioxo-3-benzazocine (23). mp >270°C; IR (KBr) 3350-3050, 1660, 1640, 1625 cm⁻¹; UV λ_{max} (log ε) 268 (4.06), 440 (2.80) nm; ¹H NMR δ 1.95 (3H, s, quinone CH₃), 2.45 (3H, s, NCH₃), 2.79 (2H, d, J = 3.0 Hz, 6-Hα and 6-Hβ), 2.90 (3H, s, NCH₃), 3.04 (1H, d, J = 12.2 Hz, 2-Hβ), 3.65 (1H, td, J = 3.0, 1.0 Hz, 5-H), 3.91 (1H, dd, J = 12.2, 5.3 Hz, 2-Hα), 3.98 (1H, dd, J = 5.3, 1.0 Hz, 1-H); MS, *m/z* (relative intensity) 276 (M⁺, 22), 217 (29), 205 (100), 204 (50), 177 (28), 176 (47). Anal. Calcd for C14H16N2O4·1/4MeOH: C, 60.20; H, 6.03; N, 9.85. Found: C, 60.54; H, 6.02; N, 9.45.

1,2,3,4,5,6,7,10-Octahydro-1,5-imino-9-methoxy-3,8,11-trimethyl-7,10-dioxo-3-benzazocine (21b). Bis(salicylidene)ethylenediiminocobalt(II) (27.0 mg, 0.083 mmol) was added to a stirred solution of 3b (37.7 mg, 0.1439 mmol) in dry DMF (4 mL). The dark suspension was stirred under an oxygen atmosphere for 5 h at room temperature. The mixture was filtered through cellulose powder and the filter cake was carefully washed with ethyl acetate. The combined filtrates were washed with water (20 mL), dried, and concentrated in vacuo to give the residue (42.1 mg). This material was subjected to chromatography on a silica gel (8 g) with dichloromethane-methanol (50:1) to afford 21b (21.8 mg, 54.9 %) as a solid. This material was recrystallized from ethyl acetate-ether to give pale yellow prisms: mp 106-107 °C, whose spectra were identical with those of the authentic sample described earlier.⁴a)

Acknowledgements. We are grateful to Professor Shin-ichiro Sakai and Dr. Koreharu Ogata of Chiba University for X-ray crystallographic determination of compound 16. We thank Mr. N. Eguchi, Ms. A. Minagawa, and Ms. S. Yoshioka in the Analytical Center of our College for measurement of spectral data (NMR and MS) and microanalytical data. We are indebted to Dr. Hiromitsu Takayama for educational discussion concerning dehydroxylation of compound 16.

References and Notes.

- 1 Ikeda, Y.; Idemoto, H.; Hirayama, F.; Yamamoto, K.; Iwao, K.; Asao, T.; Munakata, T. J. Antibiot. 1983, 36, 1284-1289; Cooper, R.; Unger, S. *ibid.* 1985, 38, 24-30.
- 2 Kubo, A.; Saito, N. Studies in Natural Products Chemistry; Atta-ur-Rahman, Ed.; Elsevier: Amsterdam, 1992, Vol. 10, pp 77-145, and references therein.
- 3 Saito, N.; Obara, Y.; Azumaya, M.; Kubo, A. Chem. Pharm. Bull. 1992, 40, 2620-2626.

- (a) Saito, N.; Õhira, Y.; Kubo, A. Chem. Pharm. Bull. 1990, 38, 821-823; Saito, N.; Õhira, Y.; Wada, N.; Kubo, A. Tetrahedron 1990, 46, 7711-7728. (b) Kubo, A.; Saito, N.; Yamato, H.; Yamauchi, R.; Hiruma, K.; Inoue, S. Chem. Pharm. Bull. 1988, 36, 2607-2614; Kurihara, H.; Mishima, H. Tetrahedron Lett. 1982, 23, 3639-3640.
- IUPAC name and numbering of tricyclic compounds are used in this paper. 5
- Plug, J. P. M.; Koomen, G. -J.; Pandit, U. K. Synthesis 1992, 1221-1222.
- Saito, H.; Kobayashi, S.; Uosaki, Y.; Sato, Y.; Fujimoto, K.; Miyoshi, K.; Ashizawa, T.; Morimoto, M.; 7 Hirata, T. Chem. Pharm. Bull. 1990, 38, 1278-1285.
- Matsumoto, M.; Kobayashi, H.; Hotta, Y. J. Org. Chem. 1984, 49, 4740-4741.
- The Baever-Villiger oxidation of 6 using Matsumoto protocol at room temperature for 92 h failed, an acetal (24) was obtained in 82.2% yield. ¹H NMR δ 2.25 (3H, s, ArCH₃), 2.44 (3H, s, NCH₃), 2.81 (1H, d, J = 16.8 Hz, 6-H β), 2.83 (3H, s, NCH₃), 3.16 (1H, dd, J = 16.8, 7.3 Hz, 6-H α), 3.31 (3H, s, OCH₃), 3.36 (1H, d, J = 12.2 Hz, 2-H β), 3.53 (3H, s, OCH₃), 3.58 (1H, d, J = 7.3 Hz, 5-H), 3.71 (3H, s, OCH3), 3.82 (1H, dd, J = 12.2, 5.6 Hz, 2-H α), 4.86 (1H, d, J = 5.6 Hz, 1-H), 5.64 (1H, s, CH), 6.89 (1H, s, 7-H); ¹³C NMR δ 15.9 (q), 27.7 (t), 33.7 (q), 40.0 (q), 51.7 (d), 54.9 (t), 55.1 (q), 56.5 (q), 58.6 (d), 61.6 (q), 103.2 (d), 128.3 (s), 128.9 (s), 129.6 (s), 132.2 (s), 133.3 (d), 155.8 (s), 170.5 (s). Treatment of 24 with 30% H2O2, H2SO4 in methanol under reflux for 24 h gave the phenol (3a) and the ester (9) in 17% and 44% yields, respectively.



- 10 Buck, P.; Köbrich, G. Tetrahedron Lett. 1967, 1563-1565; Buck, P; Köbrich, G. Chem. Ber. 1970, 103, 1412-1419.
- 11 Wiriyachitra, P.; Cava, M. P.; J. Org. Chem. 1977, 42, 2274-2277.
- 12 Regioselective functionalization of bridgehead carbanion in bicyclic piperazinedione was discussed by Williams in the chemistry of bicyclomycin synthesis; see: Williams, R. M.; Dung, J. -S.; Josey, J.; Armstrong, R. W.; Meyers, H. J. Am. Chem. Soc. 1983, 105, 3214-3220; Williams, R. M.; Durham, C. A. Chem. Rev. 1988, 88, 511-540.
- 13 Acetylation of the phenols (3b) and (12) with acetic anhydride in pyridine afforded the acetates (13) and (14) in 51% and 90% yields, respectively. The amine (4b) was identical with those of an authentic sample obtained from 4a (AIH3, THF) in 86% yield; see: Experimental section.
- 14 McOmie, J. F. W.; Watts, M. L.; West, D. E. Tetrahedron 1968, 24, 2289-2292; Bhatt, M. V.; Kulkarni, S. U. Synthesis 1983, 249-282.
- 15 Cacchi, S.; Ciattini, P. G.; Morera, E.; Ortar, G. Tetrahedron Lett. 1986, 27, 5541-5544.
- 16 The O-demethylation of 19 employing several reagents (e.g., iodotrimethylsilane^{17a}), sodium cvanidedimethyl sulfoxide^{17b}), sodium N-methylanilide in hexamethylphosphoric triamide^{17c}), iodotrichlorosilane^{17d}) failed; only the starting material was recovered.
- Vickery, E. H.; Pahler, L. F.; Eisenbraun, E. J. J. Org. Chem. 1979, 44, 4444-4446. (b) McCarthy, J. R.; Moore, J. L.; Cregge, R. J. Tetrahedron Lett. 1978, 5183-5186. (c) Loubinoux, B.; Coudert, G.; Guillaumet, G. Synthesis 1980, 638-640. (d) Bhatt, M. V.; El-Morey, S. S. Synthesis 1982, 1048-1050.
 Ahmad, R.; Saá, J. M.; Cava, M. P. J. Org. Chem. 1977, 42, 1228-1230.
- Yoshida, K.; Nakajima, S.; Ohnuma, T.; Ban, Y.; Shibasaki, M.; Aoe, K.; Date, T. J. Org. Chem. 1988. 19 53, 5355-5359.
- 20 Cambie, R. C.; Grimsdale, A. C.; Ratledge, P. S.; Woodgate, P. D. Aust. J. Chem. 1990, 43, 485-501.

(Received in Japan 12 January 1994; accepted 4 February 1994)